### **Determination of Phenolic Compounds in wines**

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#### Abstract

Wine contains natural antioxidants such as phenolic compounds also known as bioactive compounds. Samples of commercially available Greek wines were analyzed in order to determine this phenolic content. For the analysis, Reversed Phase-High Performance Liquid Chromatography (RP-HPLC) coupled with a multiwavelength Ultraviolet/visible (UV/vis) detector was used. The most abundant phenolic substances detected were (+)-catechin (13.5-72.4 mg L<sup>-1</sup>), gallic acid (0.40-99.47 mg L<sup>-1</sup>) and caffeic acid (0.87-33.48 mg L<sup>-1</sup>). The principal component analysis (PCA) technique was used to study differentiation among wines according to their theproduction area. Red wines contained more phenolic substances than white ones. Differences of the phenolic composition in wines of the same cultivar were investigated too.

Keywords: Greek wines; phenolic compounds; RP-HPLC; PCA analysis

#### 1 Introduction

Polyphenols are secondary metabolites naturally present in wine grapes and/or produced during the wine making process. They are responsible for the colour, flavour, astringency and hardness of wines as well as for their antioxidant properties (Makris, Psarra, Kallithraka, & Kefalas, 2003; Seruga, Novak, & Jakobek, 2011; Kallithraka, Kim, Tsakiris, Paraskevopoulos, & Soleas, 2011). The latter is associated with biological effects, in particular prevention of cancer, cardiovascular disease and other degenerative diseases (Fernandez-Mar, Mateos, Garcia-Parrilla, Puertas, & Cantos-Villar, 2012). Consequently, polyphenols have received considerable attention. Additionally, phenolics act as: a) metal chelators (Hider, Liu, & Khodr, 2001),

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b) antimutagens and anticarcinogens (Ferguson, 2001; He, Sun, & Pan, 2008), c) antimicrobial agents (Nychas, Tassou, & Skandamis, 2003; Rodriguez Vaquero, Alberto, & Manca de Nadra, 2007) and d) clarifying agents (Ferreira, Picarra-Pereira, Monteiro, Loureiro, & Teixeira, 2001). They are formed from the fruit and vine stems, by the yeast metabolism and/or other raw materials from vegetal origin. In addition, phenolics serve as important oxygen reservoirs and substrates for browning reactions. Phenolic compounds are also significant in white wines, where they occur at much lower concentrations. Aging in wood barrels results in a temporary increase of the phenolic content. They are possibly extracted from wood barrel. Cask wood acts as an extraction support for various phenolic compounds (Alañon, Castro-Vazquez, Diaz-Maroto,

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Gordon, & Perez-Coello, 2011). The main aim of this study was to determine phenolic compounds in red and white wines, as well as to study differentiation among wines according to their production area by using principal component analysis (PCA) technique. Also the differences of the phenolic composition in wines of the same cultivar were studied.

#### 2 Experimental

#### 2.1 Standards and solvents

Gallic acid, p-coumaric acid, ferulic acid, syringic acid, (+)-catechin and quercetin were purchased from Sigma-Aldrich (Steinheim, Germany). Caffeic acid was obtained from Merck (Darmstadt, Germany) and (-)-Epicatechin was from Fluka AG (Buchs, Switzerland). Hydroxytyrosol was a kind donation from the National Agricultural Research Foundation (N.AG.RE.F, Greece). Stock solutions of all the standards were prepared (1000 mg L<sup>-1</sup>) in water-acetic acid-acetonitrile (62:6:32 v/v/v). Working standards were made by diluting the stock solutions in the same solvent. Both stock and working standards were stored at -18°C until needed.

### 2.2 Wine samples

A total of 24 wine samples were analyzed. Most of them were commercial wines. A list of all wines analyzed in this study is presented in Table 1. The HPLC analysis was performed without any particular treatment except filtration through membrane filters 0.45 µm (Millex-HV). Each determination was carried out in triplicate.

### 2.3 HPLC Analysis

The HPLC apparatus used for the analysis consisted of a ternary gradient unit (Jasco CG-1580-02), a pump (Jasco PU-980), a multiwavelength detector UV/vis (Jasco MD-910) programmed to take data from 250-400 nm with a 4 nm resolution, a data processing system (Jasco DP-L910/V), a reversed phase column Nova Pak $(\mathbb{R})$ , C18 with a packing 4 µm (Waters, USA) protected by a guard column Nova Pak $(\mathbb{R})$ , C18, and finally, a rheodyne injection system (model 7725i) with a loop of 20  $\mu$ L. Nova Pak( $\hat{\mathbf{R}}$ ) stationary phase, which was used in this study to separate phenolic acids and flavonoids of wines, produced satisfactory results. Gradient elution of three solvents was used: Solvent A consisted of: acetic-water (1:99 v/v), solvent B: acetic-water (6:94 v/v) and solvent C: acetic-acetonitrilewater (5:30:65 v/v/v). The gradient program used was based on that of (Garcia-Parrilla, Heredia, & Troncoso, 1999): 100% A initially, 100% B 0-15 min, 100% B 15-30 min, 90% B/10%C 30-50 min, 80% B/20% C 50-60 min, 70% B/30% C 60-80 min, 100% C 80-120 min, 100% C 120-140 min. The flow rate was  $0.5 \text{ mL min}^{-1}$  and the temperature was set at  $22.5^{\circ}$ C. The monitoring wavelength was 278 nm. The identification of each compound was based on a combination of retention time and spectral matching. Quantification was done via a calibration with standards with LOD 0.06 ppm and LOQ 0.18 ppm.

### 2.4 Statistics and data presentation

Data are reported as mean and standard deviation. The % relative standard deviation (RSD) was also determined. The mean values obtained in the different samples studied were compared by MANOVA (Multi-Factor Analysis of Variance). PCA was employed by using the software Statistica v6.0 SR, to distinguish wines based on their phenolic composition. PCA permitted us to achieve a reduction of dimensionality, exploring the relationships between objects, estimating the correlation structure of the variables and investigating how many components were necessary to explain the greater part of variance with a minimum loss of information. When PCA is performed on autoscaled matrix data the principal component loadings are eigenvectors of the correlation matrix (Wold, Esbensen, & Geladi, 1987).

#### 3 Results and Discussion

The present method is simple, easy to use and effective enough for the identification and quantification of major phenolic compounds in aro-

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Sample	Label	Color	Cultivar(s)	Location
no.				
1	Kokkineli, 1999	Red	Various	Central Greece
2	Conv. culture, 7	Red	Agiorgitiko	Peloponnese
	harrel 2000			
3	Organic culture	Red	Agiorgitiko	Pelononnese
0	2000	Itou	119101910110	releponnese
4	Strofilia, Org. Culture, 1999	White	Roditis	Peloponnese
5	Wine A, 2000	Red	Various	Kriti
6	Wine B, 1998	White	Various	Central Greece
7	Kritikos topikos	White	Vilana	Kriti
	oinos, Mpoutaris 1999			
8	Strofilia, 1997,	Red	Various	Attiki
	vine-harvest			
9	Strofilia,	White	Savvatiano	Central Greece
	convetional			
	culture, 2000			
10	Strofilia, organic	White	Savvatiano	Central Greece
	culture, 2000			
11	Strofilia, organic	White	Chardonnay	Central Greece
	culture, 2000			
12	Strofilia, organic	Red	Cabernet Sauvignon	Attica
10	culture, 2000	Dal	<b>X</b> 7	D-1
13 14	Wine Nerres 2000	Red	Various	Peloponnese
14 15	Wine of Patros	Red	Various	Delenennese
10	1999	neu	Mavrodapinn	relopointese
16	Kokkineli,	Red	Various	Attica
	Kourtakis 2000			
17	Red wine of	Red	Various	Attica
	Nemeas 2001			
18	Wine D, 2000	Red	Agiorgitiko+ Cabernet Sauvignon	Peloponnese
19	Wine E 1999	White	Various	Attica
20	Makedonikos, 1999	Red	Various	Macedonia
21	Wine F, 2000	Red	Various	Attica
22	Savvatiano, 2000	White	Savvatiano	Central Greece
23	Cellar 2001	White	Fileri+Roditis+ Chardonnay	Peloponnese
24	Agiampelos, 2000	White	Various	Peloponnese

Table 1: List of the analyzed wine samples

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Figure 1: Comparison of phenolic concentration in red and white wines.

matic plants. A similar technique or previous treatment of the sample, in other words, simple filtration has been reported by other authors, for the analysis of major phenolic compounds (Gambelli & Santaroni, 2004; Rodriguez-Bernaldo de Quiros, Lage-Yusty, & Lopez-Hernandez, 2009). The amount of phenolic compounds detected in the samples is shown in Table 2. Results are expressed in mg  $L^{-1}$  wine sample  $\pm$  standard deviation. Another phenolic compound which was detected in most of the samples was hydroxytyrosol which ranged from 0.22 to 54.27 mg  $L^{-1}$ . In the making of red wine, fermentation only takes place after the maceration phase on the whole must that is obtained after crushing, and pressing. Maceration enables the extraction of constituents present in the skins and seeds into the fermenting must, including not only polyphenols and red pigments, but also tannins, volatile compounds and aroma precursors, and plant cell wall polysaccharides. Shi, Yu, Pohorly, and Kakuda (2003) showed that the phenolic compounds commonly found in white grapes (seeds removed) are esters of hydroxycinnamic acid, catechins, and procvanidins. Phenolics in red grapes contain mainly hydroxycinnamic acid-tartaric acid esters, procyanidins, flavonol glycosides, and anthocyanins. Phenolic substances were present in higher amounts in red wines than in the white ones, probably due to the prolonged contact period of the pomace with skin. This hypothesis is supported by the comparison of the mean concentration of each phenolic substance present in the red and white wines respectively as shown in Figure 1.

# 3.1 Differences of the phenolic composition in wines of the same cultivar

Samples no 9, 10 and 22 were wines of the same cultivar (Savvatiano). For these samples, no significant differences were observed although they come from different farms. Each analysis was performed in triplicate. Statistical analysis (t-test, Statistica, 1991) showed that the differences between the wines were not significant (p > 0.05). The similarities in the phenolic composition of these three wines are shown in Figure 2. Sample no. 10 was a conventional wine and sample no. 11 was an organic culture. There were no discrete differences between the two wines that could distinguish them. Although the number of samples analyzed was very small and no conclusion can be drawn, it is believed that the concentration of trans-resveratrol (Tinttunen & Lehtonen, 2001) may offer a good criterion for distinguishing normal from organic wine.



Figure 2: Comparison of the levels of phenolics in wines of the same cultivar (Savvatiano).

# 3.2 Principal component analysis (PCA) for phenolic compounds in both red and white wines

The phenolic content and composition of wine is greatly influenced by four agroecological factors: the cultivar, the year of production (i.e.,

<b>П</b> \$Я%	Ι.	75	65	63	87	89		12	88	00	79	41	54	69	10		50	87	22	08	69	50	03	69
d d d d		2.2.	37 3.	2.2	5 3.	6 3.		39 3.	3 2.	2.2	а 3.	34 2.	37 2.	8 1.	0 2.		9 2.	41 2.	2 3.	2 2.	4 2.	54.	ы. 13	1.
диетсе́tin	ND	$4.38 \pm 0.1$	$10.11\pm0.3$	$0.76\pm0.0$	$1.29\pm0.0$	$1.54\pm0.0$	ND	$12.50\pm0.3$	$1.04\pm0.0$	$0.97\pm0.0$	$0.79 \pm 0.0$	$14.10\pm0.3$	$14.53\pm0.3$	$4.71 \pm 0.0$	$4.74 \pm 0.1$	QN	$3.60\pm0.0$	$14.24\pm0.4$	$0.62 \pm 0.0$	$5.75 \pm 0.1$	$5.19 \pm 0.1$	$1.11\pm0.0$	$0.66\pm0.0$	$0.59 \pm 0.0$
USA%	3.57	4.86	2.85	ı	3.15	ı	ı	3.70	ī	ı	4.49	2.99	2.91	2.58	3.20	ï	1.88	3.76	ī	2.72	3.38	ı	ŀ	ï
Ferulic Acid	$0.56\pm0.02$	$0.97 \pm 0.05$	$0.70 \pm 0.02$	ND	$3.17 \pm 0.10$	ND	ND	$1.08 \pm 0.04$	ND	ND	$0.89 \pm 0.04$	$1.67 \pm 0.05$	$2.06 \pm 0.06$	$1.16\pm0.03$	$1.56 \pm 0.05$	ND	$1.06\pm0.02$	$1.86 \pm 0.07$	ND	$4.03 \pm 0.11$	$1.18 \pm 0.04$	ŊŊ	ND	ND
USA%	1.87	2.35	2.26	2.46	2.19	2.85	3.08	4.74	1.38	2.29	2.15	2.77	2.74	3.50	3.62	1.58	2.21	2.15	1.61	2.34	2.61	2.19	2.35	,
biɔA эітвтиоэ-q	$1.07\pm0.02$	$4.48 \pm 0.11$	$5.75 \pm 0.13$	$1.22 \pm 0.03$	$7.75 \pm 0.17$	$0.70 \pm 0.02$	$2.27 \pm 0.07$	$4.85 \pm 0.23$	$0.72 \pm 0.01$	$0.87 \pm 0.02$	$0.93 \pm 0.02$	$3.96 \pm 0.11$	$3.28 \pm 0.09$	$4.57 \pm 0.16$	$4.69 \pm 0.17$	$0.63 \pm 0.01$	$3.61 {\pm} 0.08$	$2.32 \pm 0.05$	$1.86 \pm 0.03$	$4.27 \pm 0.10$	$4.98 \pm 0.13$	$0.91 \pm 0.02$	$0.85 \pm 0.02$	QN
MSAD	4.55	4.71	2.55	1.74	3.58	ı	0.82	2.45	ı	ı	0.67	2.54	4.16	0.97	0.39	4.09	3.58	1.49	0.81	4.05	3.63	ī	1.66	2.76
ninetechin	$13.39 \pm 0.61$	$23.5\pm1.11$	$47.42 \pm 1.21$	$2.29 \pm 0.04$	$15.89 \pm 0.57$	ND	$7.32 \pm 0.06$	$47.19{\pm}1.16$	ND	ND	$2.97 \pm 0.02$	$42.01{\pm}1.07$	$26.90{\pm}1.12$	$158.42 \pm 1.55$	$37.92 \pm 0.15$	$24.66 \pm 1.01$	$29.88 \pm 1.07$	$77.72 \pm 1.16$	$7.32 \pm 0.06$	$17.52 \pm 0.71$	$18.45 \pm 0.67$	ΠN	$1.80 \pm 0.03$	$13.40 \pm 0.37$
USA%	3.20	1.88	3.03	ı	2.50	ı	ı	1.80	3.38	4.28	3.92	3.82	2.66	2.81	2.92	2.98	3.28	3.01	ı	4.46	3.46	3.84	ŀ	,
biəA əigniryZ	$1.56\pm0.05$	$5.41 {\pm} 0.10$	$1.32 \pm 0.04$	ND	$2.00 \pm 0.05$	ND	ND	$7.75 \pm 0.14$	$0.59 \pm 0.02$	$0.70 \pm 0.03$	$0.51 {\pm} 0.02$	$3.14{\pm}0.12$	$1.88 \pm 0.05$	$2.13 \pm 0.06$	$2.05 \pm 0.06$	$3.02 \pm 0.09$	$5.17\pm0.17$	$5.30 \pm 0.16$	ND	$3.36 {\pm} 0.15$	$3.75{\pm}0.13$	$0.78 \pm 0.03$	ND	QN
%BSD	3.41	3.66	3.89	3.61	2.29	2.69	2.84	2.29	2.03	3.73	2.02	2.43	2.36	3.35	2.22	2.08	1.24	4.45	2.68	1.64	1.21	1.53	2.16	3.44
Caffeic Acid	$3.81 \pm 0.13$	$8.9 \pm 0.33$	$4.88 \pm 0.19$	$2.49\pm0.09$	$33.48 \pm 0.77$	$5.19 \pm 0.14$	$5.67 {\pm} 0.16$	$16.14 \pm 0.37$	$5.41 {\pm} 0.11$	$5.35 \pm 0.20$	$2.97 \pm 0.06$	$12.33\pm0.30$	$5.92 \pm 0.14$	$8.64{\pm}0.29$	$15.71 \pm 0.35$	$4.31 \pm 0.09$	$4.01 \pm 0.05$	$7.18 \pm 0.32$	$6.34{\pm}0.17$	$23.14 \pm 0.38$	$9.05{\pm}0.11$	$4.57 \pm 0.07$	$6.46 \pm 0.14$	$0.87 \pm 0.03$
MSAN	4.16	3.53	1.79	2.77	3.92	4.54	ī	1.88	ī	ī	2.23	1.85	3.25	3.23	1.55	3.86	1.71	1.27	1.28	2.89	3.36	ī	3.48	1.74
Hydroxytyrosol	$0.24 \pm 0.01$	$24.0\pm 0.85$	$50.68 \pm 0.91$	$1.80 \pm 0.05$	$21.89 \pm 0.86$	$0.22 \pm 0.01$	ND	$46.14 \pm 0.87$	ND	ND	$3.13 \pm 0.07$	$48.44 \pm 0.90$	$25.78 \pm 0.84$	$25.99 \pm 0.84$	$5.15 \pm 0.08$	$14.50 \pm 0.56$	$54.27\pm0.93$	$8.63 {\pm} 0.11$	$10.09 \pm 0.13$	$27.26 {\pm} 0.79$	$22.88 {\pm} 0.77$	ND	$13.49\pm0.47$	$6.88 \pm 0.12$
MSSD %	4.12	3.57	4.15	4.16	4.39	4.11	4.97	3.17	4.40	4.43	4.46	2.57	3.49	3.60	3.15	4.03	2.47	0.57	3.98	3.40	0.12	4.25	3.43	4.44
тіпээtвэ-(+)	$15.26 \pm 0.63$	$28.22 \pm 1.01$	$40.20 \pm 1.67$	$16.13 \pm 0.68$	$22.30 \pm 0.98$	$24.05 \pm 0.99$	$17.49\pm0.87$	$57.04{\pm}1.81$	$15.67 \pm 0.69$	$16.92 \pm 0.75$	$20.60 \pm 0.92$	$46.21{\pm}1.19$	$24.31 \pm 0.85$	$21.89 \pm 0.79$	$9.82 \pm 0.31$	$21.08\pm0.85$	$72.44{\pm}1.79$	$57.05 \pm 1.33$	$19.84 \pm 0.79$	$24.73 \pm 0.84$	$39.91{\pm}1.05$	$17.85\pm0.76$	$27.10\pm0.93$	$16.18 \pm 0.72$
MSSD			ı	ı		1.38	1.06		3.57	2.56	1.16	ī	,	,	2.85	,	ī	ī	ī	ī	,	1.96	ī	,
p-hydroxy benzoic Acid	ΠŊ	ND	ND	ND	ND	$0.72 \pm 0.01$	$0.94 \pm 0.01$	ND	$0.38 \pm 0.01$	$0.39 \pm 0.01$	$0.86 \pm 0.01$	ND	ND	ND	$1.05 \pm 0.03$	QN	ND	ND	ND	ND	ND	$0.51 \pm 0.01$	ND	Q
%BSD	0.78	1.27	1.15	ı	1.19	ı	1.55	0.98	ı	3.57	·	1.02	0.94	1.05	1.04	0.94	1.40	1.07	2.5	1.19	1.41	2.50	·	4.00
Gallic Acid	$19.18\pm0.15$	$28.98 \pm 0.37$	$44.15 \pm 0.51$	ND	$35.10 \pm 0.42$	ND	$1.29 \pm 0.02$	$33.55 \pm 0.33$	ND	$0.56 \pm 0.02$	ND	$33.30 \pm 0.34$	$31.75 \pm 0.30$	$71.77 \pm 0.75$	$12.48 \pm 0.13$	$16.88 \pm 0.16$	$26.35 \pm 0.37$	$99.47 \pm 1.07$	$3.02 \pm 0.08$	$29.36 \pm 0.35$	$22.65 \pm 0.32$	$0.40 \pm 0.01$	ND	$1.25 \pm 0.05$
Sample no.	-	2	ŝ	4	2	9	7	x	6	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24

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the climatic condition from year to year), the site of production (the effect of geographic origin of grapes, soil chemistry, and fertilization), and the degree of maturation (Shi et al., 2003). Since the polyphenolic content varies from one area to another, it has been used in an attempt to distinguish the wines according to production area, as well as differences between red and white wines. As regards white wines, only two principal components were used whose eigenvalues were greater than one (eigenvalue-greater-thanone rule in the case of standardized data). In the case of the red wines, three principal components were used. The principal components scores for both the red and white wines (Figure 3) showed that the wines from central Greece to a greater extent and those from Attica to a lesser extent, were located inside specific regions of factors 1 and 2 and thus, there was a relationship between the compounds quantified and the aforementioned wines. Factor 1 was mainly dependent on (+)-catechin, (-)-epicatechin, caffeic acid and quercetin whereas factor 2 depended on gallic and p-coumaric acids. It was shown clearly that only for the wines coming from the Peloponnese was a relationship between the phenolic substances used and their origin. It was found that all the wines from Peloponnese are concentrated all together (Figures 4, 5 and 6). Factor 1 mainly depended on (+)-catechin and quercetin, factor 2 depended on the hydroxycinnamic acids used and finally, factor 3 was dependent on (-)-epicatechin. A similar PCA analysis was used to determine the most important differences in phenolic content among wines from the Canary Islands (Spain) according to different categories such as island, zone and denomination of origin (Rodriguez-Delgado, Gonzalez-Hernandez, Conde-Gonzalez, & Perez-Trujillo, 2002). In general, red wines from the Canary Islands had a polyphenol content in the lower part of the range considered normal. The exception was quercetin, with a mean content higher than in other wines, which may be a peculiarity of these particular wines.

### 4 Conclusion

Phenolic compounds constitute a diverse group of secondary metabolites which are present in both grapes and wines. The results of this study have shown that red wines contained higher amounts of phenolic substances than white ones. The most abundant phenolic substances detected in the wines were (+)-catechin (13.5-72.4 mg  $L^{-1}$ ), gallic acid (0.40-99.47 mg  $L^{-1}$ ) and caffeic acid  $(0.87-33.48 \text{ mg L}^{-1})$ . Differences in the phenolic composition in wines of the same cultivar were not found to be significant. All the samples were injected directly into the HPLC system in order to avoid possible changes. Hence, quantification was more reliable as there are no losses due to sample treatment. PCA statistical method was employed to find out if it is possible to distinguish wines based on their phenolic composition. Only wines of Central Greece were located inside specific regions of the factors 1 and 2. Factor 1 was mainly dependent on (+)-catechin, (-)-epicatechin, caffeic acid and quercetin whereas factor 2 depended on gallic and p-coumaric acids. Also wines with Peloponnese origin were the only ones which could be estimated through their phenolic content. In all the three plots (Fig. 6-8) all wines from Peloponnese were close together. Factor 1 mainly depended on (+)-catechin and quercetin, factor 2 depended on the hydroxycinnamic acids used and finally, factor 3 was dependent on (-)-epicatechin. Of course if a larger number of wines from different regions of Greece were analyzed, then it would be easier to create a reliable method for estimating the origin of wines.

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Figure 3: Plot of the two principal components scores issued from PCA for white wines (attica, kriti, pnnese=peloponnese and c.Greece=central Greece). Factor 1 is mainly dependent on (+)-catechin, (-)-epicatechin, caffeic acid and quercetin whereas factor 2 depends on gallic and p-coumaric acid.



Figure 4: Plot of the factor 1 vs. factor 2 for the red wines (santo=the greek island Santorini). Factor 1 mainly depends on (+)-catechin and quercetin, factor 2 depends on the hydroxycinnamic acids used and finally, factor 3 is dependent on (-)-epicatechin.

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Figure 5: Plot of the factor 1 vs. factor 3 for the red wines. Factor 1 mainly depends on (+)-catechin and quercetin, factor 2 depends on the hydroxycinnamic acids used and finally, factor 3 is dependent on (-)-epicatechin.



Figure 6: Plot of the factor 2 vs. factor 3 for the red wines. Factor 1 mainly depends on (+)-catechin and quercetin, factor 2 depends on the hydroxycinnamic acids used and finally, factor 3 is dependent on (-) epicatechin.

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